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METAL – TITANIUM DIOXIDE DOPED CATALYSTS FOR WASTEWATER TREATMENT UNDER SIMULATED SOLAR LIGHT

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Introduction

TiO₂ assisted photo catalysis was widely studied in the last years for the advanced degradation of organic compounds from aqueous systems, mainly for those not removed by conventional treatment processes. The main drawback of TiO₂ is that it is activated only by UV radiation and uses only about 5% of available solar radiation. TiO₂ doping with metals proved to be a good method to improve its photocatalytic properties and to enhance its response to solar radiation.

Materials and methods

Three types of metal-doped TiO₂ catalysts (Fe-TiO₂, Ni-TiO₂, Co-TiO₂) were prepared using sol-gel method. The TiO₂ precursor was titanium (IV) isopropoxide (Sigma Aldrich) and Fe(NO₃)₃ · 9 H₂O (Sigma Aldrich), Ni(NO₃)₂ · 6 H₂O (Sigma Aldrich), Co(NO₃)₂ · 6 H₂O (Sigma Aldrich) were used as metal sources. Catalysts with 1wt%, 2wt%, 5wt% were prepared. A lamp with the following characteristics: 35W, 380-800 nm, 1000 lumens, photosynthetic photon flux density 300 μmol quanta/m²·s was used to simulate solar radiation. Dimensional analyse was performed using a Mastersizer 2000 – Malvern equipment and a FEI Quanta FEG 250 scanning electronic microscope was used for morphological characterisation and EDX (energy dispersive X – ray spectroscopy) characterisation. Two sets of experiments were performed using a synthetic solution of methylene blue - MB (Merck) and real wastewater from a municipal wastewater treatment plant.

Results and conclusions

Prepared catalysts were first characterized from the point of view of particle dimensions. The obtained results showed the following dimensional parameters: Fe-TiO₂ [d (0.1) = 0.388 μm; d(0.5) = 0.505 μm; d(0.9) = 0.979 μm]; Ni- TiO₂ [d (0.1) = 0.176 μm; d(0.5) = 0.331 μm; d(0.9) = 0.634 μm]; Co- TiO₂ [d (0.1) = 0.176 μm; d(0.5) = 0.330 μm; d(0.9) = 0.630 μm]. Morphological characterisation confirmed the results of dimensional analyses showing that Ni-TiO₂ and Co-TiO₂ exhibits smaller dimensions compared with Fe-TiO₂. Moreover, SEM images (Figure 1) showed that particles present non-regular shapes with the dimensions in the domain of hundreds of nanometres. EDX analyses confirmed the dopants presence within catalysts structure (Table 1).

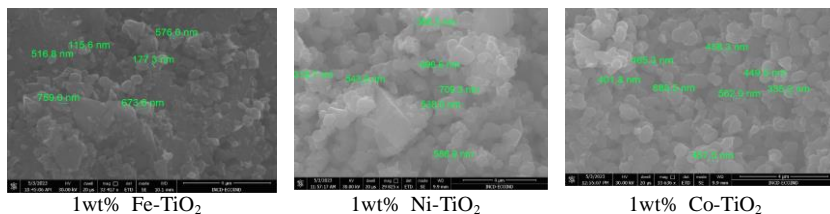
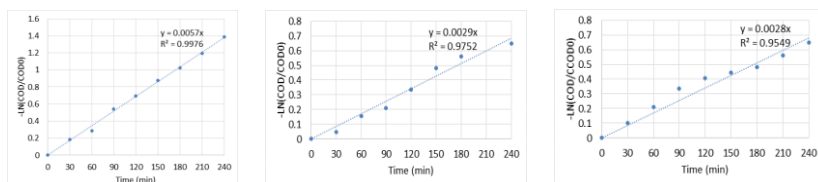


Figure 1. SEM images of metal doped titanium dioxide catalysts

Table 1. EDX results

1wt% Fe-TiO ₂		1wt% Ni-TiO ₂		1wt% Co-TiO ₂	
Element	Atomic %	Element	Atomic %	Element	Atomic %
O	66.43	O	66.97	O	70.53
Fe	0.22	Ni	0.21	Co	0.90
Ti	33.35	Ti	32.82	Ti	28.57

Photocatalytic activity of all catalysts were tested using a MB (as model organic compound) solution. MB initial concentration and after 3 hours of exposure to solar simulated radiation and a catalyst dose of 100 mg/L were measured based on the absorbance at 662 nm (corresponding to maximum MB absorbance). Best results were obtained for 2wt% Metal – TiO₂ catalysts with a MB degradation efficiency of 6.00% for 2wt% Co-TiO₂, 6.91% for 2wt% Ni-TiO₂ and 13.82% for 2wt% Fe-TiO₂. The 2wt% Metal-TiO₂ catalysts were further used for treatment of real wastewater samples varying the initial catalyst concentration. The best degradation efficiencies for organic compounds (expressed as chemical oxygen demand - COD) after 120 minutes of irradiation were obtained for the following catalysts doses: 200 mg/L for 2wt% Ni-TiO₂ (28.57%), 100 mg/L for 2wt% Co-TiO₂ (33.33%) and 100 mg/L for 2wt% Fe-TiO₂ (50.00%). Further increase of irradiation time to 240 minutes for optimum catalysts doses led to a degradation of organic compounds (expressed as COD) of 47.69% for both 2wt% Ni-TiO₂ and 2wt% Co-TiO₂ and 75.00% for 2wt% Fe-TiO₂. This behaviour is sustained also by the linearized pseudo-first order kinetic profiles for the optimum catalysts type and doses (Figure 2).



100 mg/L; 2wt% Fe-TiO₂

200 mg/L; 2wt% Ni-TiO₂

100 mg/L; 2wt% Co-TiO₂

Figure 2. Pseudo-first order kinetic for optimum catalysts, irradiation time 240 min

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