

# Comparison of Three Metal Doped Titanium Dioxide Photo-Catalysts Performances for Treatment of Municipal Wastewater Under Simulated Solarlight

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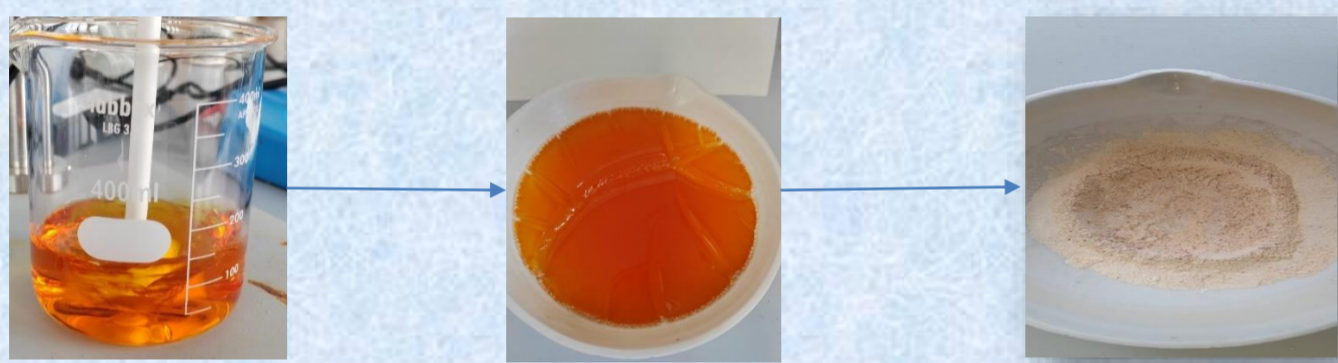


## Introduction:

This study aimed to assess performance of a solar slurry photocatalytic membrane reactor for municipal wastewater treatment, using three metal – doped TiO<sub>2</sub> catalysts (for photocatalytic step) and a polymer based membrane (for membrane separation step).

## Results & Discussions:

- Catalysts were characterized with spectrometer XRF Rigaku which confirmed the presence of Ti=98% (TiO<sub>2</sub>) and aprox. 1% metal content (Fe=0,91%, Co=1%, Ni=0,93%) also SEM analysis.
- The efficiency of organic matter removal was measured with COD parameter after 7h radiation (photocatalytic step): only radiation=16,67%, commercial TiO<sub>2</sub>=25%, synthesized TiO<sub>2</sub> = 28,57%; 1% Fe-TiO<sub>2</sub> = 61,11%, 1% Co-TiO<sub>2</sub> = 47,37%, 1% Ni-TiO<sub>2</sub> = 42,11%.
- Overall COD removal after membrane separation step: the degradation efficiency followed the order 1%Fe-TiO<sub>2</sub> > 1% Ni-TiO<sub>2</sub> > 1% Co-TiO<sub>2</sub> (residual COD = 17,6 / 61,6 / 79,2 mg O<sub>2</sub>/L).



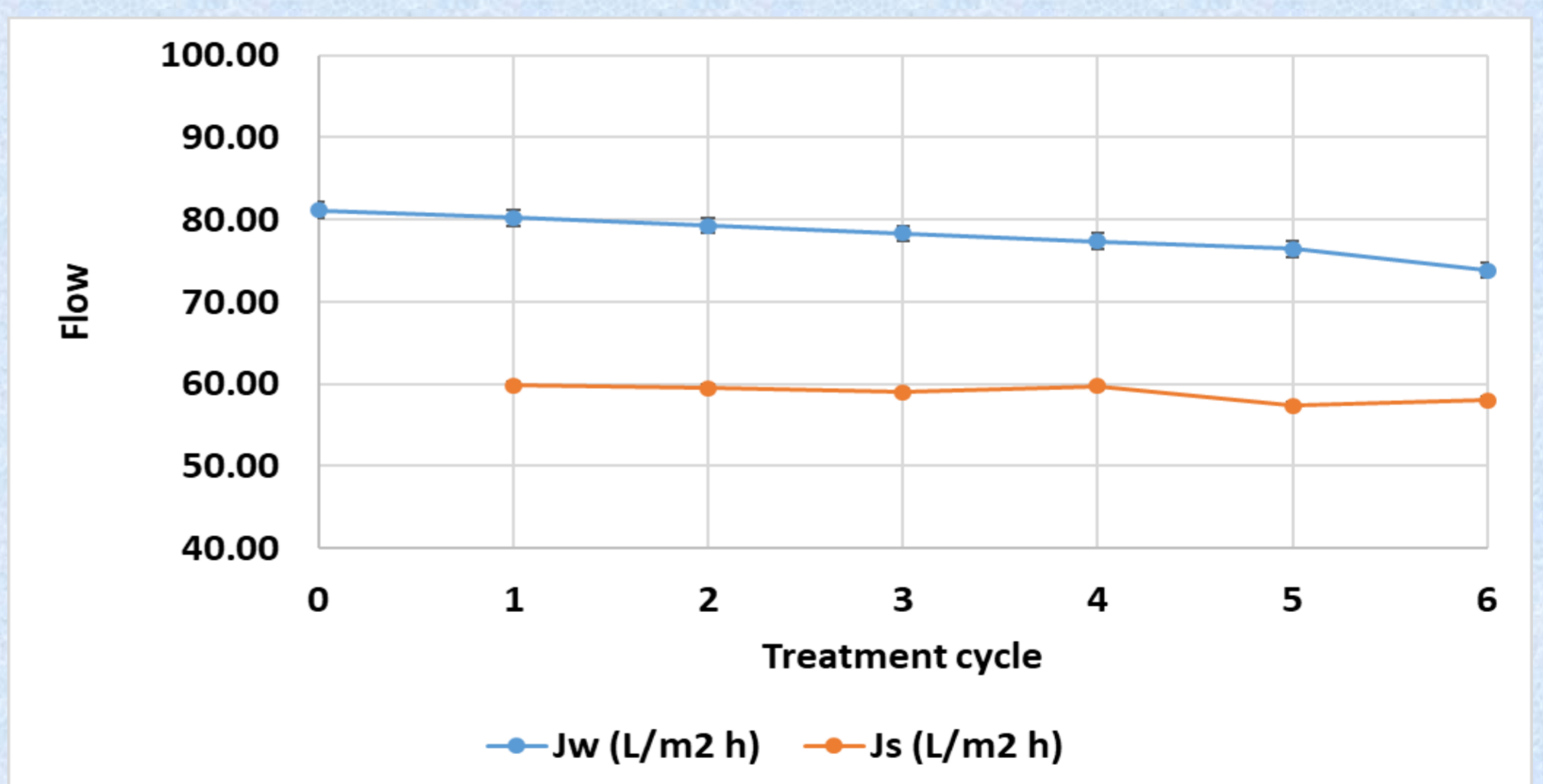
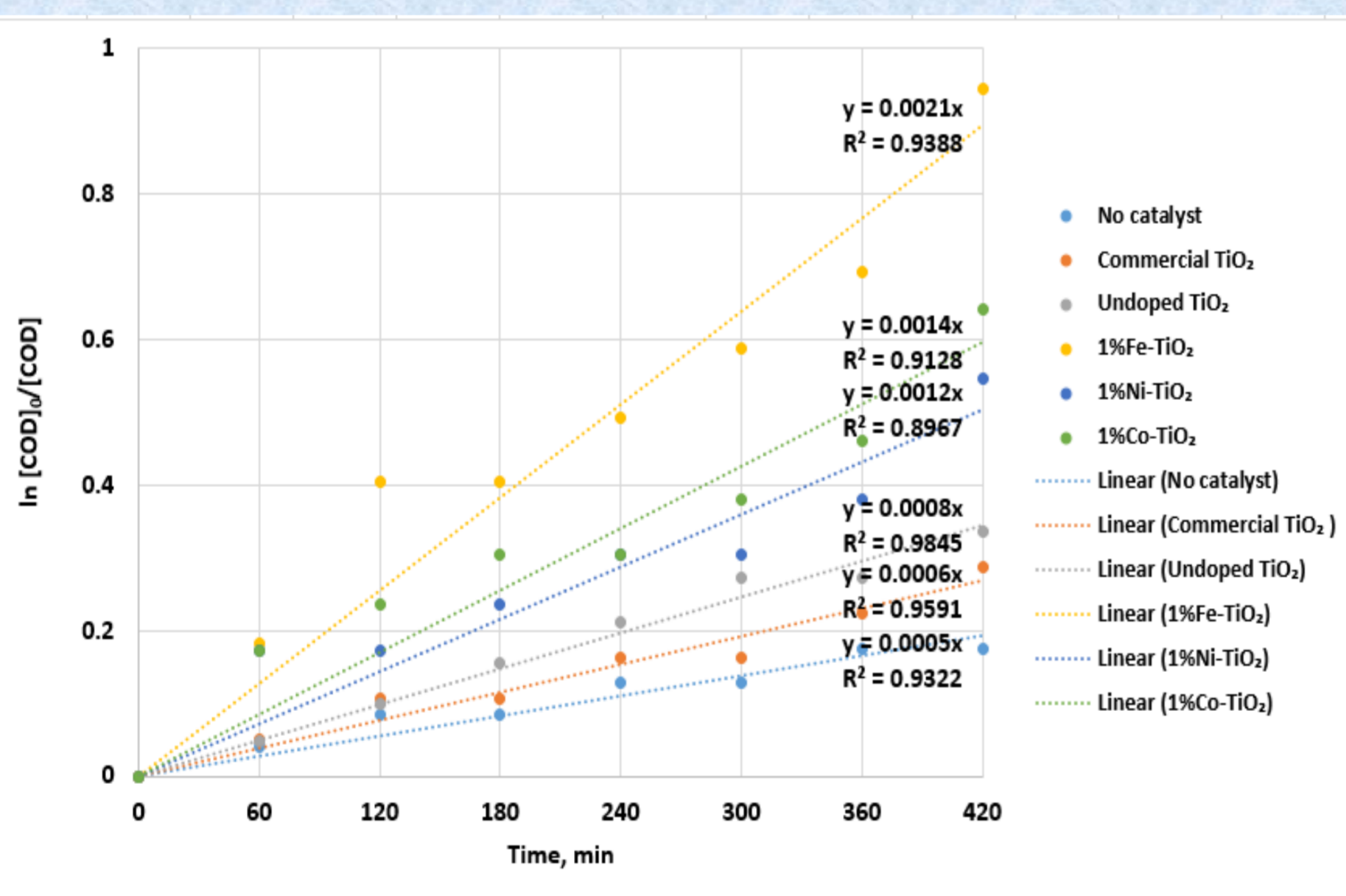
The degradation of organic compounds for photocatalytic step follows a pseudo 1st order kinetics (COD).

## Research:

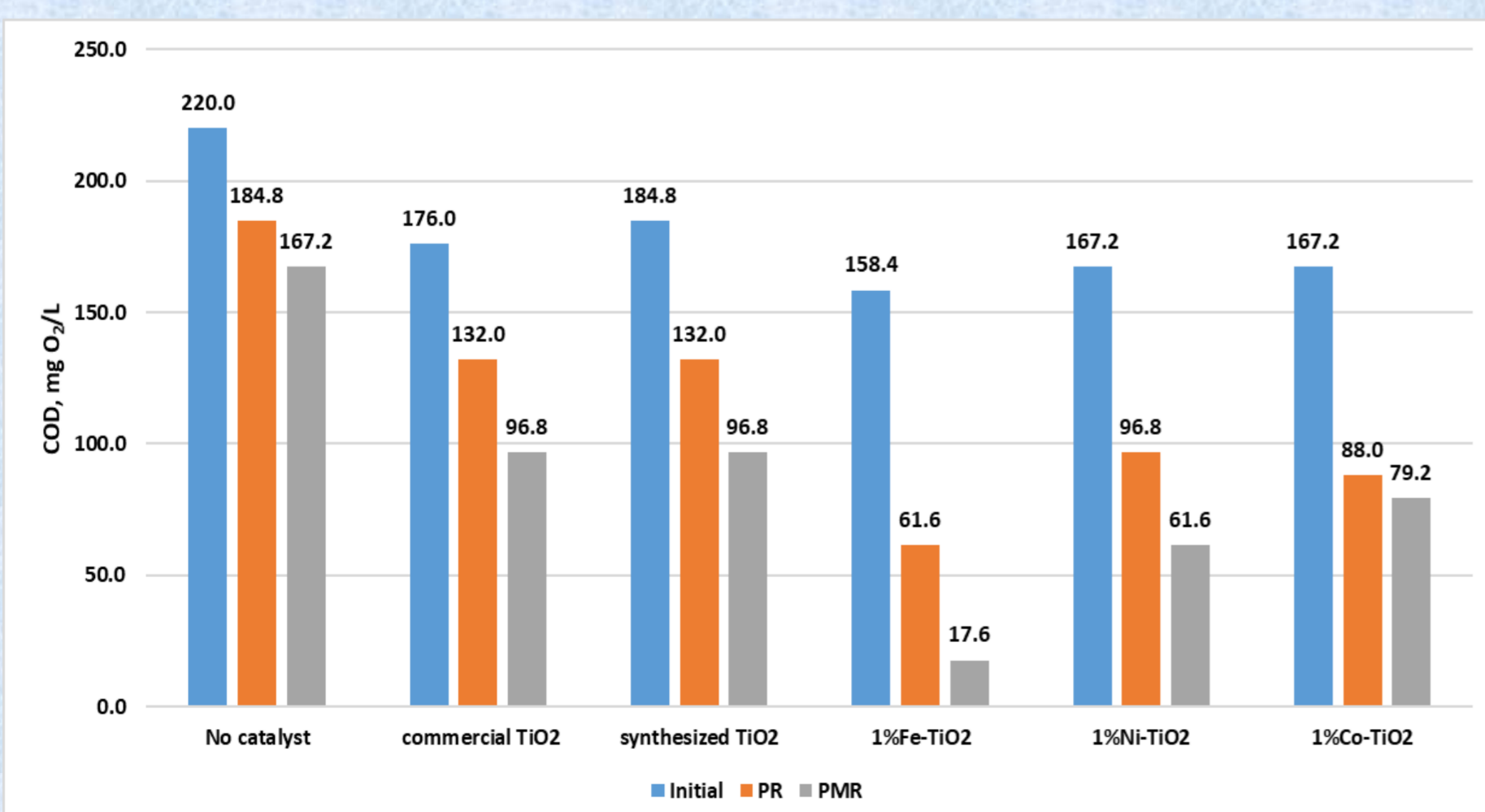
- Three different metal doped catalysts were synthesized by sol-gel method (1% Fe-TiO<sub>2</sub>, 1% Ni-TiO<sub>2</sub> and 1% Co-TiO<sub>2</sub>) and compared with commercial TiO<sub>2</sub> – anatase form (Merck) and un-doped TiO<sub>2</sub>.
- The concentration of catalyst used in experiments was 200 mg/L and each wastewater sample was exposed to radiation for 7 hours by simulated solar light (intensity within 35-47 W/m<sup>2</sup>).
- A polymeric membrane was synthesized using a solution of 12% Psf (polysulfone) by phase inversion technique and used after the irradiation process to recover the catalyst.
- A custom made photocatalytic installation was used for the experimental part & composed of 4 PETG transparent tubes, feed/recirculation pump connector, supply/recirculation pump, aeration system, plastic panel support (covered with Al) and a supply/recirculation tank (2L).
- For the separation & recycling stage was used KMS Laboratory Cell – CF 2 (Koch Membrane Systems) (max. 500 mL volume, working pressure 2.5-3 bar & 28 cm<sup>2</sup> effective surface area ).



After 6 separation cycles the membrane didn't show any significant changes in the separation flow and ultrapure water flow.



Fe doped TiO<sub>2</sub> catalyst presented best removal efficiency (COD=17,6 mg O<sub>2</sub>/L, efficiency=88,89%) while TSM were less than 5mg/L for every experiment:



## CONCLUSION

- The removal efficiency of the organic matter (COD) varied depending on used catalyst as follows: 1%Fe-TiO<sub>2</sub> > 1% Ni-TiO<sub>2</sub> > 1% Co-TiO<sub>2</sub> > synthesized TiO<sub>2</sub> > commercial TiO<sub>2</sub> > only radiation.
- Doped catalysts increase the efficiency in removing organics from water compared to the un-doped TiO<sub>2</sub>: 88,89% (1% Fe-TiO<sub>2</sub>), 63,16% (1% Ni-TiO<sub>2</sub>) and 52,63% (1% Co-TiO<sub>2</sub>) versus 47,62% (TiO<sub>2</sub> synthesized) and 45,00% (commercial TiO<sub>2</sub>).
- By using the catalyst 1% Fe-TiO<sub>2</sub> the COD was reduced from 158,4 to 17,6 mg O<sub>2</sub>/L, obtaining the highest efficiency (88,89%).
- The experiments highlighted that the solar PMR technological flow is a viable alternative for real wastewater treatment.
- Membrane fouling was evaluated – after 6 treatment cycles the membrane system maintains its efficiency: reduction of only 4,01% for separation flow & 9,07% for ultrapure water flow.

## Acknowledgements

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