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SCAN-RATE INFLUENCE ON SENSING PROPERTIES OF AN ETHANOL ELECTROCHEMICAL SENSOR BASED ON PANI-TiO₂ NANOCOMPOSITES ACTIVATED WITH GOLD NANOPARTICLE

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Introduction

The quality control of alcoholic beverages during the fermentation process is influenced by the measurement of ethanol level which necessitates having rigorous analytical methods for ethanol in beverages. For this purpose, an electrochemical sensor based on a screen-printed carbon electrode (SPCE) was developed modified with TiO₂/PANi nanocomposites, and activated with gold nanoparticles (AuNP). TiO₂ nanotube array is a promising functional material for applications in gas sensors and biosensors due to its flexibility in adjusting the microstructure in terms of tube length, tube diameter and chemical composition, highly uniform morphology, and large surface area with controllable pore sizes.

Materials and methods

A screen-printed carbon microelectrode (SPCE) with a hybrid configuration of carbon/ceramic nanotube/conductive polymer/metallic nanoparticle (CNF/TiO₂/PANi/AuNP) is developed on a standard 3-electrode electrochemical sensor structure which is coated by screen printing methods with a thin layer of carbon nanofiber with DRP-CNF SOL ink (Dropsens), carbon nanofibers with a diameter of 40-80 nm and a length of 0.5 - 1.5 μm with a helical structure. Initially, very thin polymer fibers were obtained from polymethyl methacrylate (PMMA) by electrospinning. The polymeric nanofibers were coated with TiO₂ ceramic oxides by sputtering, respectively magnetic spraying with a radio frequency magnetron, the polymeric core being removed by heat treatment which implicitly led to the obtaining of TiO₂ ceramic nanotubes. TiO₂ ceramic nanotubes were coated with conductive PANi polymers by drop-casting and then deposited on the SPCE. Finally, the sensor was activated with Au particles by drop-casting and evaporation.

Results and conclusions

From scanning electron microscopy (SEM) and atomic force microscopy (AFM) analysis images (Figure 1) it can be seen how the smaller gold nanoparticles attach themselves to the oxide nanotubes or probably fill in the empty spaces.

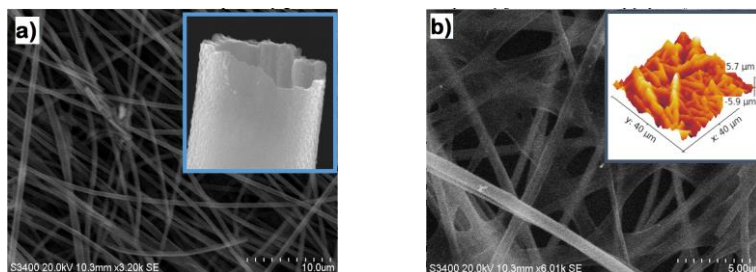


Fig. 1. SEM image TiO₂ nanotubes+PANi (inset TiO₂ nanotubes) si b) SEM image TiO₂ nanotubes+PANi+AuNPs (inset 3D AFM)

The purpose of the tests was to determine the influence of the scan rate and the potential domain on the sensitivity of the hybrid electrode to ethanol (concentrations of 9.5, 14, and 24 ppm). The measurements were made at two potential ranges: -0.1 – 1.5 V și ± 1 V and at two scan rates of 30 mV/s and 90 mV/s, respectively. The best results to determine the concentration of ethanol were obtained at the scan rate of 30 mV/s in the potential range of -0.5 V ÷ 1.5 V (Figure 2).

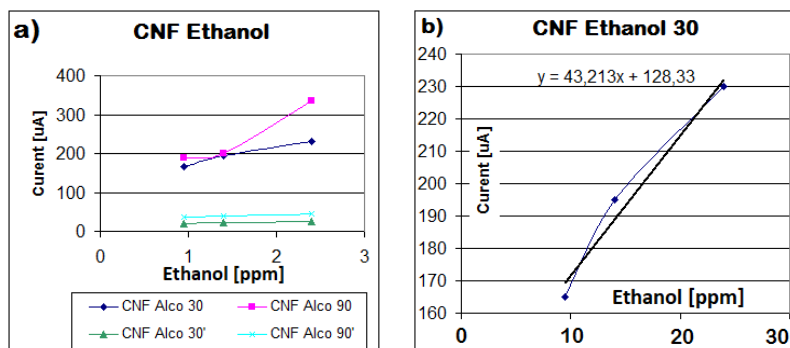


Fig. 2. a) The comparative diagram with the sensor's response to ethanol b) the diagram of the oxidation peak values at 30mV scan rate and -0.5 V ÷ 1.5 V potential domains.

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